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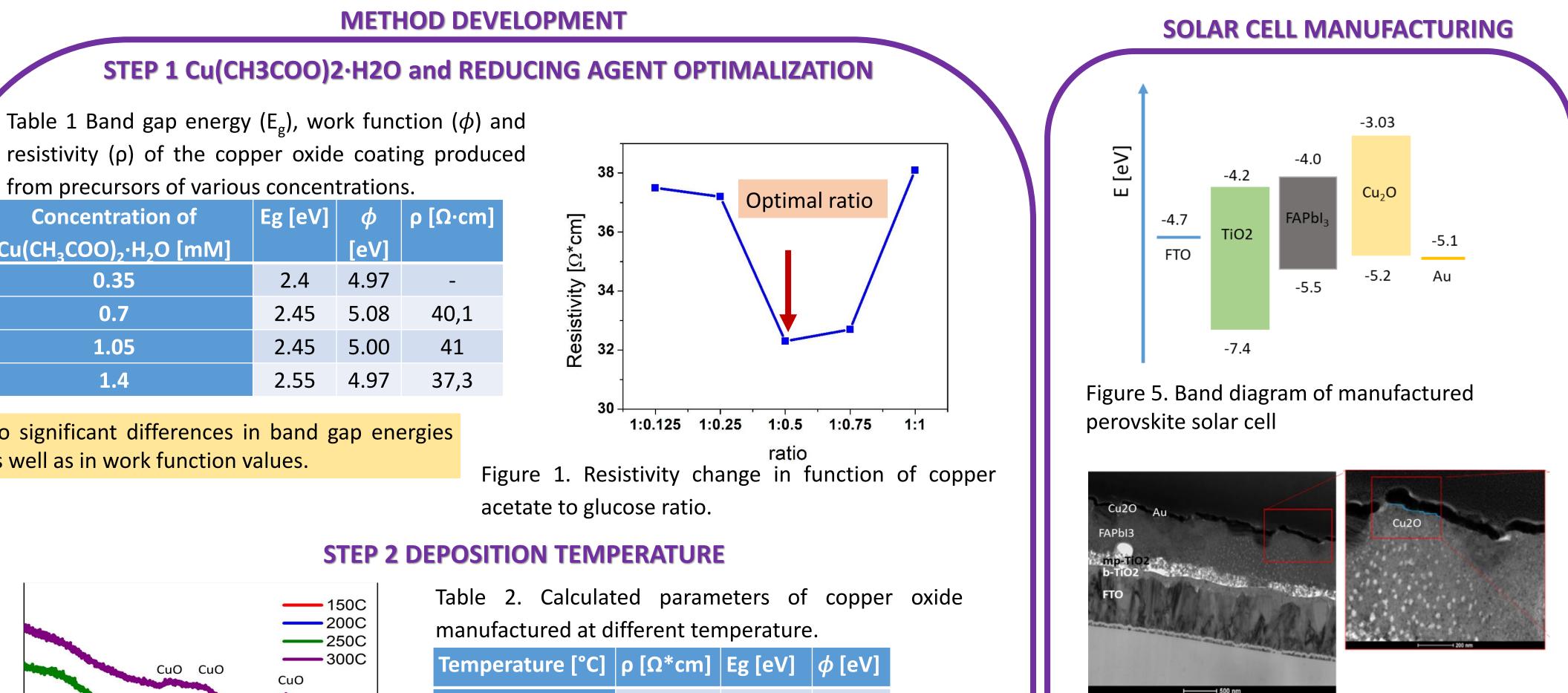


COPPER (I) OXIDE THIN FILM PREPARED BY SPRAY COATING METHOD

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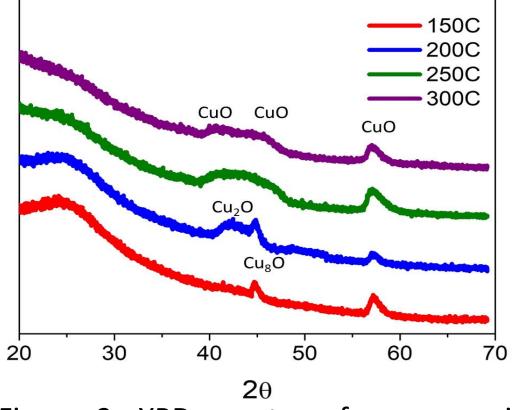
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In this work, the copper oxide was manufactured by a simple and scalable spray coating technique from copper acetate solution in air. The investigation included the optimal concentration of copper acetate monohydrate and glucose finding and the determination of beneficial deposition temperature. What is more, to obtain single phase Cu2O the thermal post treatment at 300°C in an inert atmosphere of nitrogen was implemented. Due to that the resistivity of the layer was reduced from 23.2 Ω*cm to 21 Ω*cm. Also, the transformation of CuO in as deposited layer to Cu₂O after additional treatment was confirmed by XRD analysis. The grain size calculated by Scherrer equation increased from 6.17 nm to 10.42 nm. The charge carrier concentration and mobility for pure Cu₂O were 2.99e+16 1/cm3 and 3.44 cm2/(V*s) respectively. It should be highlighted that this measurement was not possible for a deposited layer. In addition, the first attempt to use copper oxide in perovskite solar cells as hole transporting material (HTM) was made. Due to the gentle nature of perovskite, the thermal Cu₂O was deposited on top of it with thicknesses of 4.5 nm, 9 nm, and 18 nm. The results revealed that the use of Cu₂O improves the electrical parameters of perovskite solar cells compared to the reference sample without HTM. However, obtained IV parameters are much lower than that for devices with Spiro-OMeTAD.



$Cu(Cn_3COO)_2 \cdot n_2O[mivi]$		[ev]	
0.35	2.4	4.97	-
0.7	2.45	5.08	40,1
1.05	2.45	5.00	41
1.4	2.55	4.97	37,3

No significant differences in band gap energies as well as in work function values.



Temperature [°C]	ρ [Ω*cm]	Eg [eV]	φ [eV]
150	-	2,5	-
200	30	2,5	5,16
250	40.9	2,5	5,08
300	37.4	2	5,12

Figure 2. XRD spectra of copper oxide layer manufactured at different temperatures

Different layer composition depending on the temperature. At 150°C the coating is a mixture of a CuO (57°) and not stable Cu8O (44.5°). With temperature increasing the Cu8O peak disappears. Cu2O (131) peak at 42° shows up at 200°C, it is broadened at 250°C and finally at 300°C only three CuO peaks are visible 41.5°, 45° and 57°.

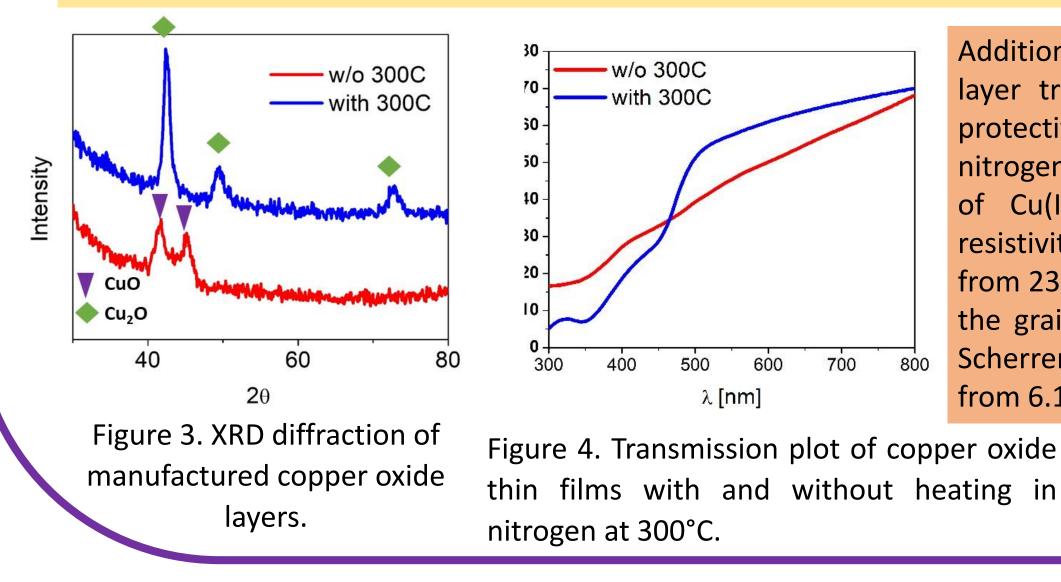
STEP 3 HIGH TEMPERATURE POST TREATMENT

Figure. 6 TEM cross-section of a perovskite cell with thin Cu₂O layer

 Table 3. I-V parameters of perovskite solar cells
with copper oxide as HTL

	Jsc [mA/cm²]	Voc [mV]	FF [-]	η [%]
Ref.	13.4	315	0.31	1.30
Spiro	25.58	966.99	0.654	16.18
4.5 nm Cu ₂ O	17.83	654.01	0.508	5.91

Additional high temperature post treatment at 300°C in an inert atmosphere of nitrogen.



Additional high temperature layer treatment at 300°C in a atmosphere protective of nitrogen allows for the reduction of Cu(II) to Cu(I), the layer resistivity slightly decreased from 23.2 Ω^* cm to 21 Ω^* cm and the grain size calculated by the Scherrer equation increased from 6.17 to 10.42 nm.

9 nm	17.8	805	0.38	5.4
Cu ₂ O	17.0	605	0.56	5.4
18 nm	12.7	612	0.43	3.38
Cu ₂ O	12.7	012	0.45	5.50

The results indicated that reference perovskite solar cell with no HTM has worse IV parameters than those with Cu₂O HTM. It was found that for the thinner copper oxide the better electrical parameters are observed.

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